

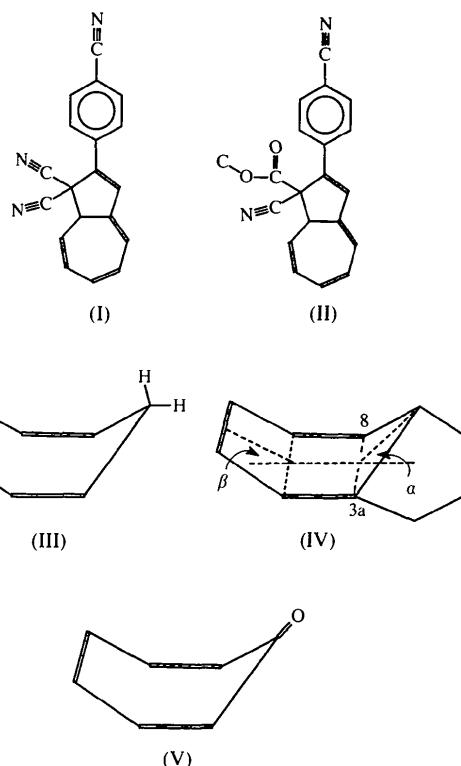
1993). Diffraction des cristaux insuffisante pour obtenir plus de réflexions observées et correction d'absorption négligée.

Des documents complémentaires concernant cette structure peuvent être obtenus à partir des archives électroniques de l'UICr (Référence: PA1250). Les processus d'accès à ces archives est donné au dos de la couverture.

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1990; Daub, Fischer, Salbeck & Ulrich, 1990; Achatz, Fischer, Salbeck & Daub, 1991). The crystal structures of few 1,8a-dihydroazulenes are known (see, for example, Brown, McPhail & Sim, 1966; Van de Grampel, Cuperus & Vos, 1971; Daub *et al.*, 1985, 1986).



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## 2-*p*-Cyanophenyl-1,8a-dihydroazulene-1,1-dicarbonitrile and Methyl 1-Cyano-2-*p*-cyanophenyl-1,8a-dihydroazulene-1-carboxylate

MENAHEM KAFTORY,<sup>a</sup> MARK BOTOSHANSKY,<sup>a</sup> JÖRG DAUB,<sup>b</sup> CHRISTIAN FISCHER<sup>b</sup> AND ALEXANDER BROSS<sup>b</sup>

<sup>a</sup>Department of Chemistry, Technion-Israel Institute of Technology, Haifa, Israel 32000, and <sup>b</sup>Institut für Organische Chemie, Universität Regensburg, Universitätsstraße, D-8400 Regensburg, Germany. E-mail: kaftory@tx.technion.ac.il

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### Abstract

The title compounds,  $C_{19}H_{11}N_3$  and  $C_{20}H_{14}N_2O_2$ , show solid-state photochromic properties. In both compounds, the dihydroazulene moiety was found to adopt the boat conformation.

### Comment

Our interest in 1,8a-dihydroazulenes stems from their use as chemical and physical switches caused by a photochemically induced rearrangement (see, for example, Daub *et al.*, 1989; Daub, Gierisch & Salbeck,

Crystals of 2-*p*-cyanophenyl-1,8a-dihydroazulene-1,1-dicarbonitrile, (I), were poorly diffracting and there was a complete fall off of intensity above  $\theta = 22.0^\circ$ . There is a clear distinction between  $C_{sp^2}$ — $C_{sp^2}$  double and single bonds in the cycloheptatriene moiety. The formal double-bond lengths (Table 1) are in the range 1.338 (5)–1.345 (3) Å and the formal  $C_{sp^2}$ — $C_{sp^2}$  single-bond lengths are in the range 1.434 (3)–1.441 (5) Å. The five-membered rings are almost planar [deviations from the mean planes are smaller than 0.077 (2) Å]. Irradiation of the photochromic compounds (I) and (II) (methyl 1-cyano-2-*p*-cyanophenyl-1,8a-dihydroazulene-1-carboxylate) results in an opening of the fused cyclopentene moiety. The expected consequence of the ring opening is the conformational change of the cycloheptatriene moiety. It is therefore interesting to note the expected differences of the conformation in both cases by comparing the conformation of cycloheptatriene in three systems: cycloheptatriene, (III), as the reference system, dihydroazulene, (IV), and heptafulvene, (V). The cycloheptatriene in all three systems adopts the boat conformation. A comparison of the three important parameters that describe the conformation is given in Table 2. The notation is shown in (IV) [definition of the  $\alpha$  and  $\beta$  angles were taken from Daub *et al.*

(1985)]. There is a very small increase of the C3a···C8 distances in dihydroazulene attributed to the flattening of the cycloheptatriene moiety and expressed by the decrease of the  $\beta$  angle. Further flattening is observed in heptafulvene and it is shown by the decrease of both the  $\alpha$  and  $\beta$  angles, followed by an increase of the C3a···C8 distances. The conclusion is that the cycloheptatriene in both photochromic compounds (I) and (II) is flattened during the irradiation.

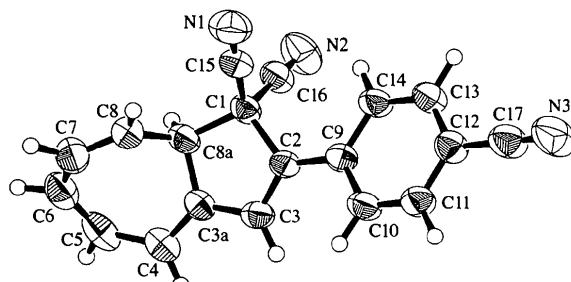


Fig. 1. The molecular structure of (I). The displacement ellipsoids are drawn at the 50% probability level.

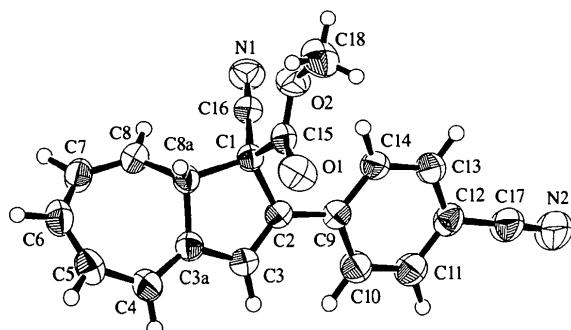


Fig. 2. The molecular structure of (II). The displacement ellipsoids are drawn at the 50% probability level.

## Experimental

The procedure for the synthesis of similar compounds is given in Daub *et al.* (1986).

### Compound (I)

#### Crystal data

C<sub>19</sub>H<sub>11</sub>N<sub>3</sub>  
 $M_r = 281.31$   
Triclinic  
 $P\bar{1}$   
 $a = 10.016(4)\text{ \AA}$   
 $b = 9.185(4)\text{ \AA}$   
 $c = 9.105(4)\text{ \AA}$   
 $\alpha = 102.68(4)^\circ$   
 $\beta = 116.90(5)^\circ$   
 $\gamma = 84.62(3)^\circ$   
 $V = 728.8(5)\text{ \AA}^3$   
 $Z = 2$   
 $D_x = 1.282\text{ Mg m}^{-3}$   
 $D_m$  not measured

Mo K $\alpha$  radiation  
 $\lambda = 0.71069\text{ \AA}$   
Cell parameters from 25 reflections  
 $\theta = 3.1\text{--}15.2^\circ$   
 $\mu = 0.078\text{ mm}^{-1}$   
 $T = 293(2)\text{ K}$   
Plate  
 $0.30 \times 0.27 \times 0.15\text{ mm}$   
Light yellow

### Data collection

Phillips PW1100 four-circle diffractometer  
 $\omega/2\theta$  scans  
Absorption correction: none  
1788 measured reflections  
1788 independent reflections  
1308 reflections with  $I > 2\sigma(I)$

$\theta_{\max} = 22^\circ$   
 $h = -10 \rightarrow 9$   
 $k = -9 \rightarrow 9$   
 $l = 0 \rightarrow 9$   
3 standard reflections  
frequency: 120 min  
intensity decay: 1.5%

### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.051$   
 $wR(F^2) = 0.136$   
 $S = 1.085$   
1788 reflections  
244 parameters  
All H atoms refined  
 $w = 1/[\sigma^2(F_o^2) + (0.065P)^2 + 0.0635P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.013$

$\Delta\rho_{\max} = 0.245\text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.139\text{ e \AA}^{-3}$   
Extinction correction:  
SHELXL93 (Sheldrick, 1993)  
Extinction coefficient:  
0.013 (6)  
Scattering factors from International Tables for Crystallography (Vol. C)

### Compound (II)

#### Crystal data

C<sub>20</sub>H<sub>14</sub>N<sub>2</sub>O<sub>2</sub>  
 $M_r = 314.33$   
Monoclinic  
 $C2/c$   
 $a = 17.965(2)\text{ \AA}$   
 $b = 13.649(2)\text{ \AA}$   
 $c = 16.987(2)\text{ \AA}$   
 $\beta = 129.514(5)^\circ$   
 $V = 3213.4(7)\text{ \AA}^3$   
 $Z = 8$   
 $D_x = 1.299\text{ Mg m}^{-3}$   
 $D_m$  not measured

Cu K $\alpha$  radiation  
 $\lambda = 1.54178\text{ \AA}$   
Cell parameters from 24 reflections  
 $\theta = 43.4\text{--}44.9^\circ$   
 $\mu = 0.687\text{ mm}^{-1}$   
 $T = 293(2)\text{ K}$   
Prism  
 $0.50 \times 0.20 \times 0.15\text{ mm}$   
Light yellow

### Data collection

Rigaku AFC-6S diffractometer  
 $\omega/2\theta$  scans  
Absorption correction: none  
3212 measured reflections  
3212 independent reflections  
2448 reflections with  $I > 2\sigma(I)$

$\theta_{\max} = 77.60^\circ$   
 $h = 0 \rightarrow 22$   
 $k = 0 \rightarrow 17$   
 $l = -20 \rightarrow 16$   
3 standard reflections every 200 reflections  
intensity decay: 0.9%

### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.038$   
 $wR(F^2) = 0.119$   
 $S = 1.099$   
3212 reflections  
274 parameters  
All H atoms refined  
 $w = 1/[\sigma^2(F_o^2) + (0.0514P)^2 + 0.9474P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.082$   
 $\Delta\rho_{\max} = 0.218\text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.198\text{ e \AA}^{-3}$   
Extinction correction:  
SHELXL93  
Extinction coefficient:  
0.00148 (14)  
Scattering factors from International Tables for Crystallography (Vol. C)

**Table 1. Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ) for (I) and (II)**

	(I)	(II)
C1—C8a	1.571 (4)	1.587 (2)
C1—C15	1.477 (5)	1.538 (2)
C1—C16	1.482 (4)	1.474 (2)
C2—C9	1.458 (4)	1.467 (2)
C1—C2	1.545 (4)	1.540 (2)
C2—C3	1.349 (4)	1.340 (2)
C3—C3a	1.432 (4)	1.433 (2)
C4—C3a	1.344 (4)	1.345 (2)
C4—C5	1.439 (5)	1.438 (2)
C5—C6	1.338 (5)	1.345 (3)
C6—C7	1.441 (5)	1.434 (3)
C7—C8	1.338 (5)	1.342 (2)
C8—C8a	1.497 (4)	1.505 (2)
C8a—C3a	1.506 (4)	1.514 (2)
C2—C1—C8a	104.3 (2)	103.8 (1)
C1—C2—C3	108.5 (3)	109.7 (1)
C2—C3—C3a	113.4 (3)	113.3 (1)
C3—C3a—C8a	108.7 (2)	108.8 (1)
C4—C3a—C8a	122.1 (3)	123.8 (2)
C3a—C4—C5	124.4 (3)	124.9 (2)
C4—C5—C6	125.7 (4)	125.6 (2)
C5—C6—C7	125.9 (4)	125.9 (2)
C6—C7—C8	125.4 (4)	127.4 (2)
C7—C8—C8a	120.7 (3)	121.3 (2)
C8—C8a—C3a	107.9 (2)	107.8 (1)
C1—C8a—C3a	103.4 (2)	103.3 (1)

**Table 2. Comparison of cycloheptatriene conformation**

REFCODE†	$\alpha$ ( $^\circ$ )‡	$\beta$ ( $^\circ$ )‡	C3a—C8a ( $\text{\AA}$ )	Reference
<b>System (III)</b>				
DMDPCH	52.6	34.3	2.410 (7)	Stegemann & Lindner (1979)
HEHWIA	48.8	28.1	2.387 (4)	Burnett <i>et al.</i> (1994)
<b>System (IV)</b>				
(I)	53.5 (3)	27.3 (2)	2.428 (4)	Present work
(II)	50.9 (2)	25.16 (9)	2.439 (2)	Present work
CRCAAZ	50.4	6.6	2.39 (5)	
	58.5	9.5	2.39 (5)	Brown <i>et al.</i> (1966)
HPHAZO	47.2	24.5	2.453 (8)	Van de Grampel <i>et al.</i> (1971)
FAPMOY	53.1	26.4	2.417 (6)	Daub <i>et al.</i> (1986)
DELCUS	53.3	27.6	2.436 (4)	Daub <i>et al.</i> (1985)
<b>System (V)</b>				
BETBEH	8.2	0.9	2.499 (2)	Reichardt <i>et al.</i> (1985)
HEPFUL	24.7	15.7	2.492 (5)	Thomas & Coppens (1972)
MCHFUL	45.3	20.1	2.474 (7)	Shimanouchi <i>et al.</i> (1974)
TAFFAH	36.6	28.9	2.461 (4)	Badejo <i>et al.</i> (1990)

† The refcodes are those used by Cambridge Structural Database (Allen & Kennard, 1993). ‡ S.u. values for  $\alpha$  and  $\beta$  angles are not given in the original articles.

Data collection: *PW1100/20 Software* (Phillips, 1973) for (I); *Rigaku MSC/AFCS Software* (Rigaku Corporation, 1988) for (II). Cell refinement: *PW1100/20 Software* for (I); *Rigaku MSC/AFCS Software* for (II). Data reduction: *PROCN* in *PW1100/20 Software* for (I); *Rigaku MSC/AFCS Software* for (II). For both compounds, program(s) used to solve structures: *SHELXS86* (Sheldrick, 1990); program(s) used to refine structures: *SHELXL93* (Sheldrick, 1993); molecular graphics: *ORTEPII* (Johnson, 1976) and *TEXSAN* (Molecular Structure Corporation, 1993); software used to prepare material for publication: WORD6.0.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: BS1031). Services for accessing these data are described at the back of the journal.

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## N-3-Allylation of 2-(*N,N*-Dimethylamino)methyleneamino)-6-formylpteridin-4-one

MADELEINE HELLIWELL, ANDREW DINSMORE, C. DAVID GARNER AND JOHN A. JOULE

*Chemistry Department, The University of Manchester, Manchester M13 9PL, England. E-mail: j.a.joule@man.ac.uk*

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## Abstract

The reaction of 2-(*N,N*-dimethylaminomethyleneamino)-6-formylpteridin-4-one with allyl bromide in the presence of diazabicyclo[4.3.0]nonane as base leads